

## STOVE MODEL II

Stove model II was made keeping the first model as the reference. This new model has a concentrator disc of dia. 11.5 cm (0.7 dia. of the combustor) and riser of diameter 15.5 cm/ height 8.5 cm from the concentrator disc. Also for supporting a pot support or even the vessel itself a cover was made with outer chamber of the previous stove model.

Part	Diameter/ Breadth (cm)	Height (cm)	Volume
Outer Shell	23.5 x 23.5 square	32	15 litres
Combustion chamber	16.5 dia.	22	5 litres
Concentrator disc	11.5 dia. (hole)	NA	NA
Riser	15.5 dia.	8.5	NA
Cover	23 x 23 square	8.5	NA

Air Inlet	Dimensions	Area
Primary air inlet area	1 cm dia. hole x 25	19.64 sq.cm
Secondary air inlet area	0.8 cm dia hole x (38+37)	37.699 sq.cm

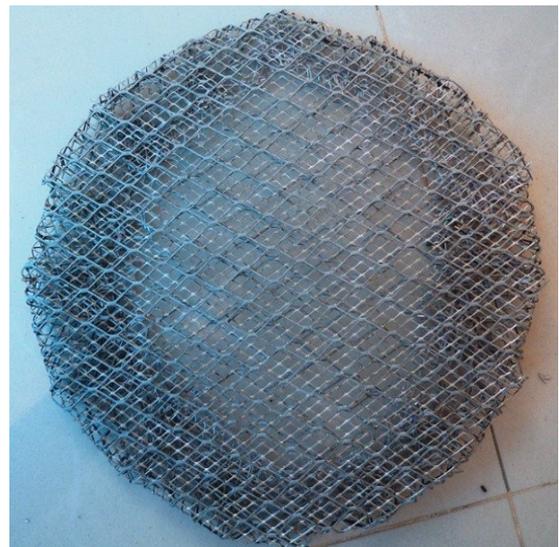
Secondary holes are kept about 4 cm below the concentrator disc in order reduce leakage problem found in the first one.

### PHOTOS:

Fig. Combustion chamber with riser



Fig. Grate



**Grate** was made from scrap of a mosquito bat.

Fig. Stove without grate and cover



Fig. Inside the stove



Concentrator disc is actually the top of the metal can (combustion chamber)

Fig. Parts of the stove

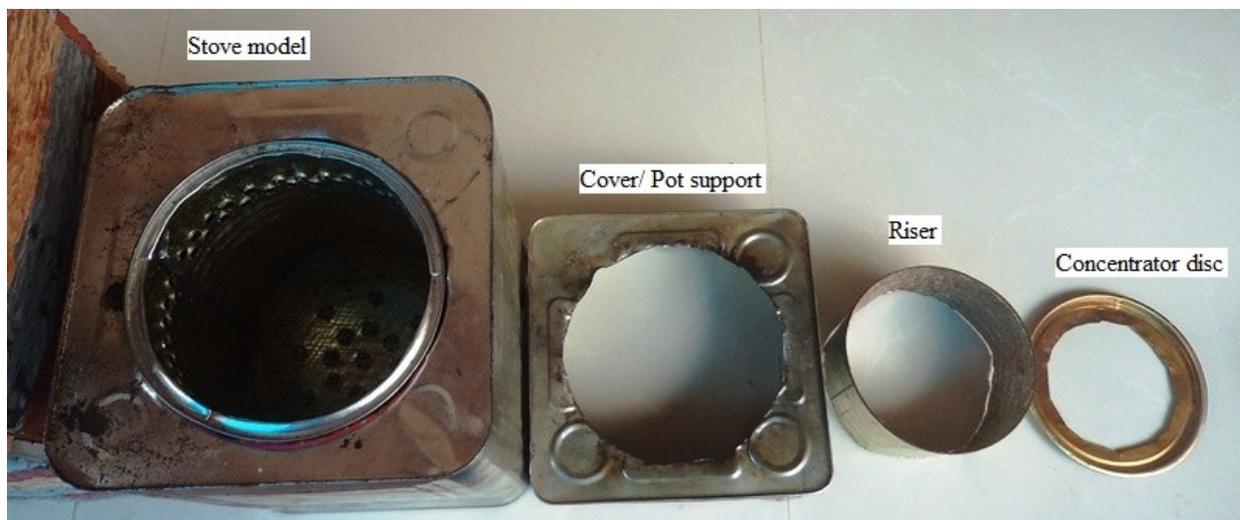


Fig. Stove model II (different views)



## Testing on stove model II

### Water Boiling Test

Method used: High Power Test –*COLD START*

Efficiency test was done on smokeless portable metallic chulha fabricated By SALOOP .T. S of Model Engineering College Trikkakkara, Ernakulam as part of his Project Work. Test was conducted at Integrated Rural Technology Centre (IRTC), Mundur, Palakkad under the supervision of Prof. B. M. Musthafa, Research Co-ordinator and assisted by Sri. Rangaswamy, Technical assistant.

Test Date **02.12.2014**

### Procedure:

The evaluation was done by conducting a water boiling test on the chulha which was brought at the centre (IRTC). The test conducted in the following manner.

- Known mass of firewood DRY COCONUT SHELL ( $M_f$ ) was taken in to lumps.
- Known mass of water ( $M_{w1}$ ) was taken in a vessel (The vessel selected so that it will rest on the stand over the stove)
- Initial temperature of water was taken ( $T_1$ )
- The stove was loaded with first lot of coconut shell and the fuel was burned.
- Fuel was loaded as and when the first lot were consumed for to maintain uniform rate of burning
- Temperature of boiling water was taken ( $T_2$ )
- When the burning of firewood came to an end the vessel was removed and the mass of water remaining measured ( $M_{w2}$ )

Mass of water evaporated =  $M_{ev} = M_{w1} - M_{w2}$

### The efficiency of stove was calculated as follows:

*Input energy* = Mass of fuel loaded x Calorific value of fuel.

*Useful output* = (Energy used to heat water from initial temperature up to stage of boiling + energy used for quantity of water evaporated)

= (Mass of water taken x Sensible heat x differences in temperature) + (Mass of water evaporated x Latent heat of water)

*Percentage of efficiency* = (Useful energy output/Energy input) x 100

$$= [M_{w1} \times S_w (T_2 - T_1) + M_{wev} \times L] / [M_f \times C_{vf}] \times 100$$

$S_w$  - Sensible heat of water (Specific heat) = 1 kcal/kg  
 $L$  - Latent heat of vaporization of water = 540 kcal/kg  
 $Cv_f$  - Calorific value of coconut shell used = 5000 kcal/kg

### TEST-1

Observations:

#### 1. Firewood details:

A mixture of size 2 cm to 3 cm

Weight = 0.75 kg

#### 2. Vessel details:

Type : Steel

Diameter : 29 cm

Height : 14 cm

#### 3. Temperatures:

Room temperature : 26.5 °C

Temperature of water before heating : 25.7 °C

Temperature of boiling water : 99.3 °C

#### 4. Other details:

Test started at 12.30 pm

At 1.10 pm firewood completely burned. Remaining water measured at 1.34 pm.

Temperature of water then 80°C.

#### 5. Flame details:

Yellow

#### 6. Smoke details:

Initial stage - Light Black

After 10 minutes - No visible smoke

After 20 minutes - No visible smoke

### Efficiency calculations:

Mass of firewood loaded,  $M_f = 0.75$  kg

Calorific value of fuel,  $C_f = 5000$  kcal/kg

Initial temperature of water,  $T_1 = 25.7$  °C

Mass of water taken,  $M_{w1} = 3$  kg

Sensible heat of water (Specific heat),  $S_w = 1$  kcal/kg

Temperature of boiling water,  $T_2 = 99.3$  °C

Mass of water remaining,  $M_{w2} = 1.9$  kg

Mass of water evaporated,  $M_{W1}-M_{W2} = 3 - 1.9 = 1.1 \text{ kg}$

Latent heat of evaporation of water,  $L = 540 \text{ kcal/kg}$

$$\begin{aligned} \text{\% of efficiency} &= (\text{Useful energy output} / \text{Energy input}) \times 100 \\ &= \{3 \times 1(99.3-25.7) + (1.1 \times 540)\} / (0.75 \times 5000) \times 100 \\ &= \mathbf{21.73 \text{ \%}} \end{aligned}$$

Time in minutes	Temperature ( $^{\circ}\text{C}$ )
0	25.7
5	84.6
10	97.2
20	98.6
30	99.3
40	80 (fuel burnt completely)

Time in minute	Side surface-Bottom ( $^{\circ}\text{C}$ )	Side surface – top ( $^{\circ}\text{C}$ )
0	25.7	25.7
10	65	115
20	78.6	164.2
30	78	164
40	84	120

## TEST-2

### 1. Firewood details:

A mixture of size 2 cm to 3cm”

Weight - 0.75 kg

### 2. Vessel details:

Type: Steel

Diameter: 29 cm

Total height: 14 cm

### 3. Temperatures:

Room temperature:  $27.5^{\circ}\text{C}$

Temperature of water before heating:  $26.3^{\circ}\text{C}$

Temperature of boiling water: 99.5 °C

4. Other details:

Test started at 3:19 pm. At 3:40pm firewood completely burned.  
Remaining water measured at 3:50pm. Temperature of water then 75°C

5. Flame details:

Yellow

6. Smoke:

Initial stage - Light Black

After 10 minutes - No visible smoke

After 20 minutes - No visible smoke

**Efficiency calculations:**

Mass of firewood loaded,  $M_f = 0.75$  kg

Calorific value of fuel,  $C_f = 5000$  kcal/kg

Initial temperature of water,  $T_1 = 26.3$  °C

Mass of water taken,  $M_{W1} = 3$  kg

Sensible heat of water (Specific heat),  $S_w = 1$  kcal/kg

Temperature of boiling water,  $T_2 = 99.5$  °C

Mass of water remaining,  $M_{W2} = 1.944$  kg

Mass of water evaporated,  $M_{W1} - M_{W2} = 3 - 1.944 = 1.056$  kg

Latent heat of evaporation of water,  $L = 540$  kcal/kg

% of efficiency = (Useful energy output / Energy input) x 100

$$= \{3 \times 1(99.5 - 26.3) + (1.056 \times 540)\} / (0.75 \times 5000) \times 100$$
$$= \mathbf{21.06 \%}$$

Time in minutes	Temperature (°C)
0	26.3
1	31.6
5	75.8
8	99.4
13	99.5
16	99.7 (fuel burnt completely)

Time in minute	Side surface-Bottom (°C)	Side surface – top (°C)
0	25.7	25.7
1	49	63.5
4	49.7	106
5	76.5	108
13	93.4	112.5
16	99.2	108.5

**Comments and suggestions:**

- The efficiency of the Stove is lower when comparing with similar type of firewood stoves.
- The thermal energy loss from the metallic surface is too high (at the top surface temp. is >100 ° C).
- Continuous fuel feeding is not possible in the model.
- The outer case can be replaced with cast iron or terracotta.
- The position of air hole for secondary burning can be lowered to a suitable depth.
- Easy fuel feeding position and suitable holes should be separately found out in the mould.

**SALOOP T S.**

M.Tech. Energy Management, Govt. Model Engineering College, Ernakulam, Kerala ,  
India

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